The Equilibrium Distribution and its Properties

In a previous lecture we derived the H theorem for a single species only, but it can be shown that the derivation also holds for mixtures of (non-reactive) species, for which the appropriate definition is,

$$H = \sum_{species} n_j \left\langle \ln f_j \right\rangle_j = \sum_{species} \left(\int f_j \ln f_j d^3 w \right)$$

If an equilibrium is reached (dH/dt = 0), H will then have attained its minimum value, consistent with the constraints that are implicit to the H theorem. These constraints are

- (a) Conservation of the number of particles of each species (p.u. volume).
- (b) Conservation of the <u>overall</u> momentum density (but species can exchange momentum, so it is not conserved species by species).
- (c) Conservation of <u>overall</u> kinetic energy density (again, not for each species).

We therefore impose the following constraints:

$$E = \sum_{i} \int \frac{1}{2} m_{i} w^{2} f_{i}(\vec{w}) d^{3}w \quad \text{(one equation)}$$

$$\vec{P} = \sum_{i} \int m_{i} \vec{w} f_{i}(\vec{w}) d^{3}w \quad \text{(three equations)}$$

$$n_{i} = \int f_{i}(\vec{w}) d^{3}w \quad \text{(one equation per species)}$$

We adjoin Lagrange multipliers and minimize the functional:

$$\sum_{i} \int f_{i}(\ln f_{i})d^{3}w + \sum_{i} \alpha_{i} \int f_{i}d^{3}w + \beta \sum_{i} \int \frac{1}{2}m_{i}w^{2}f_{i}d^{3}w + \vec{\gamma} \cdot \sum_{i} \int m_{i}\vec{w}f_{i}d^{3}w$$
$$= \sum_{i} \int f_{i}\left(\ln f_{i} + \alpha_{i} + \beta \frac{1}{2}m_{i}w^{2} + \vec{\gamma} \cdot m_{i}\vec{w}\right)d^{3}w$$

Notice that a single Lagrange multiplier β is associated with the total sum of energies, and also a single vector $\vec{\gamma}$ is associated with the total sum of momenta; this is in fact the origin of the eventual fact that $\vec{u}_i = \vec{u}$ and $T_i = T$. Differentiation relative to f_i gives then,

$$\ln f_i + \alpha_i + \beta \frac{1}{2} m_i w^2 + \vec{\gamma} \cdot m_i \vec{w} + 1 = 0$$

$$f_i = e^{-(1+\alpha_i)} e^{-m_i \vec{\gamma} \cdot \vec{w} - \beta \frac{1}{2} m_i w^2}$$

$$= e^{-(1+\alpha_i) + \frac{m_i \gamma^2}{2\beta}} e^{-\frac{m_i \beta}{2} (\vec{w} + \frac{\vec{\gamma}}{\beta})^2}$$

where we have completed the square in the exponent.

The constants α_i , $\vec{\gamma}$ and β will now be determined from the constraint equations; before doing the detailed algebra however, one can readily see that the mean velocities and the temperatures must indeed be common to all species. For species *i*,

$$\vec{u}_i = \frac{1}{n_i} \int \vec{w} f_i d^3 w = \frac{\int \vec{w} f_i d^3 w}{\int f_i d^3 w} = \frac{\int \vec{w} e^{-\frac{m_i \beta}{2} (\vec{w} + \frac{\gamma}{\beta})^2} d^3 w}{\int e^{-\frac{m_i \beta}{2} (\vec{w} + \frac{\gamma}{\beta})^2} d^3 w}$$

where the common factor $e^{-(1+\alpha_i)+\frac{m_i\gamma^2}{2\beta}}$ has been dropped in the ratio. Change variable to $\vec{\zeta} = \vec{w} + \frac{\vec{\gamma}}{\beta}$:

$$\vec{u}_i = \frac{\int \vec{\zeta} e^{-\frac{m_i\beta}{2}\zeta^2} d^3\zeta - \frac{\vec{\gamma}}{\beta} \int e^{-\frac{m_i\beta}{2}\zeta^2} d^3\zeta}{\int e^{-\frac{m_i\beta}{2}\zeta^2} d^3\zeta} = \frac{-\vec{\gamma}}{\beta}$$

since the first integration vanishes by symmetry. This result is independent of i, and so $\vec{u}_i = \vec{u}$, the same for all i.

Similarly, once we know $\vec{u} = -\frac{\vec{\gamma}}{\beta}$ (and recall $\vec{c} = \vec{w} - \vec{u}$),

$$\frac{3}{2}kT_i = \frac{1}{2}m_i \left\langle c^2 \right\rangle_i = \frac{1}{n_i} \int \frac{m_i(\vec{w} - \vec{u})^2}{2} f_i d^3 w = \frac{\frac{m_i}{2} \int (\vec{w} - \vec{u})^2 e^{-\frac{m_i \beta}{2}(\vec{w} - \vec{u})^2} d^3 u}{\int e^{-\frac{m_i \beta}{2}(\vec{w} - \vec{u})^2} d^3 w}$$

and changing now to $\vec{y} = \sqrt{\frac{m_i\beta}{2}}(\vec{w} - \vec{u}),$

$$\frac{3}{2}kT_i = \frac{\frac{1}{\beta}\int y^2 e^{-y^2} d^3y}{\int e^{-y^2} d^3y}$$

The ratio of integrals turns out to be $\frac{3}{2}$, showing that,

$$T_i = \frac{1}{k\beta}$$

again independent of i. So with,

$$\beta = \frac{1}{kT}$$
 and $\vec{\gamma} = -\frac{\vec{u}}{kT}$

we have,

$$f_i(\vec{w}) = \underbrace{e^{-(1+\alpha_i) + \frac{m_i}{2} \frac{u^2}{kT}}}_{K} e^{-\frac{m_i(\vec{w} - \vec{u})^2}{2kT}}$$

$$n_{i} = \int f_{i} d^{3}w = K \int e^{-\frac{m_{i}(\vec{w} - \vec{u})^{2}}{2kT}} d^{3}w \quad \text{with} \quad \sqrt{\frac{m_{i}}{2kT}} (\vec{w} - \vec{u}) = y$$

then,

$$n_{i} = K \int_{0}^{\infty} \left(\frac{2kT}{m_{i}}\right)^{3/2} e^{-y^{2}} 4\pi y^{2} dy$$

and noticing that,

$$y^{2} = t \qquad dy = \frac{1}{2}t^{1/2}dt$$
$$n_{i} = K\left(\frac{2kT}{m_{i}}\right)^{3/2}\frac{4\pi}{2}\underbrace{\int_{0}^{\infty}t^{1/2}e^{-t}dt}_{\Gamma(\frac{3}{2})=\frac{1}{2}\sqrt{\pi}} = K\left(\frac{2\pi kT}{m_{i}}\right)^{3/2}$$

Solving for K,

$$K = n_i \left(\frac{m_i}{2\pi kT}\right)^{3/2}$$

Therefore we find,

$$f_i(\vec{w}) = n_i \left(\frac{m_i}{2\pi kT}\right)^{3/2} e^{-\frac{m_i(\vec{w}-\vec{u})^2}{2kT}}$$

which is the Maxwellian or Equilibrium distribution function.

We could re-derive the Maxwellian limit using an alternative argument to the optimization procedure discussed above. During our discussion of the H-theorem, we obtained,

$$\frac{dH}{dt} = \frac{1}{4} \int d\Omega \iint (f'f_1' - ff_1) \ln\left(\frac{ff_1}{f'f_1'}\right) g\sigma d^3w d^3w_1 \le 0$$

and the equality (equilibrium) can only be true if,

$$f'f'_1 = ff_1$$
 for all $\vec{w}, \vec{w}_1, \vec{\Omega}$

Hence the quantity,

$$\ln f(\vec{w}) + \ln f(\vec{w}_1)$$

is <u>conserved</u> in a collision between particles with velocities $\vec{w}, \vec{w_1}$. This is an <u>additive</u> quantity. What other additive quantities are conserved? The list is short; assuming zero momentum, they are:

- (a) From number conservation, any constant quantity (the quantity 1, for instance)
- (b) From energy conservation, the quantity $\frac{1}{2}mw^2$

Hence $\ln f$ must be a linear combination of these:

$$\ln f = \ln c_1 - c_3 \frac{1}{2} m w^2$$

and therefore,

$$f = c_1 e^{-\frac{c_3}{2}mw^2}$$

If there is non-zero momentum, we should include it and write,

$$\ln f = \ln c_1 + \vec{c}_2 \cdot m\vec{w} - c_3 \frac{1}{2}mw^2$$

The values of c_1 and c_3 (and $\vec{c_2}$, if needed) come from imposing normalization such that,

$$\int f d^3 w = n \qquad \int \vec{w} f d^3 w = n \vec{u} \qquad \int \frac{1}{2} m w^2 f d^3 w = n \frac{3}{2} kT$$

The result, as with the minimization method, is,

$$f(\vec{w}) = n \left(\frac{m}{2\pi kT}\right)^{3/2} e^{-\frac{m(\vec{w}-\vec{u})^2}{2kT}}$$

This method can be generalized to multi-species situations, although in that case, since there are several kinds of collisions, there will be more than one necessary conditions like $f'f'_1 = ff_1$, and some care must be exercised with terms arising from unlike particles.

Characteristic energies and velocities for a Maxwellian distribution

In a frame in which the gas is at rest $(\vec{u} = 0)$, the mean vector velocity is zero. More generally, $\langle \vec{w} - \vec{u} \rangle_s = 0$, for any species s.

We generally define $\vec{c}_s = \vec{w} - \vec{u}_s$, the velocity of a particle with regard to the mean of its species. This is sometimes called the "diffusion velocity", but care must be taken not to confuse it with $\vec{c} = \vec{w} - \vec{u}$, where \vec{u} is the mean mass velocity of <u>all</u> the species present. We see from the definition that $\langle \vec{c}_s \rangle_s \equiv 0$, but $\langle \vec{c}_s \rangle \equiv \vec{u}_s - \vec{u}$, which, in a non-equilibrium situation, can be non-zero.

An important velocity magnitude is $\overline{c}_s \equiv \langle c_s \rangle_s$, where the <u>magnitude</u>, and not the vector, is involved. For a Maxwellian,

$$\overline{c}_s = \frac{1}{p_s} \int c_s p_s \left(\frac{m_s}{2\pi kT_s}\right)^{\frac{3}{2}} e^{-\frac{m_s c_s^2}{2kT_s}} d^3 c_s$$

Since only $|\vec{c_s}|$ appears, use spherical coordinates, where $d^3c_s=4\pi c_s^2 dc_s$

$$\overline{c}_s = \int_0^\infty c_s \left(\frac{m_s}{2\pi kT_s}\right)^{\frac{3}{2}} e^{-\frac{m_s c_s^2}{2kT_s}} 4\pi c_s^2 dc_s$$

Define,

$$x^2 = \frac{m_s c_s^2}{2kT_s} \rightarrow c_s = \left(\frac{2kT_s}{m_s}\right)^{\frac{1}{2}} x \text{ and } dc_s = \left(\frac{2kT_s}{m_s}\right)^{\frac{1}{2}} dx$$

The integral can be evaluated by changing $x^2 = t$, $x^3 dx = \frac{1}{2}t dt$, and its value is $\frac{1}{2}$. So,

$$\overline{c}_s = \frac{2}{\sqrt{\pi}} \left(\frac{2kT_s}{m_s} \right)^{\frac{1}{2}} \qquad \overline{c}_s = \sqrt{\frac{8}{\pi} \frac{kT_s}{m_s}}$$

Another important velocity is the RMS velocity, or $c_{RMS} = \sqrt{\langle c_s^2 \rangle_s}$. This can be calculated more easily, in fact, for any distribution, because,

$$\langle c_s^2 \rangle \equiv \frac{2}{m_s} \frac{3}{2} kT_s \quad \rightarrow \quad c_{RMS} = \sqrt{3 \frac{kT_s}{m_s}}$$

Sometimes the distribution of interest is where particles are classified by either velocity magnitude or by energies. Looking at the first of these, we define a different distribution (assumed to be isotropic) by,

$$h(c_s)dc_s \equiv f(c_s)d^3c_s = f(c_s)4\pi c_s^2 dc_s \quad \rightarrow \quad h = 4\pi c_s^2 f$$

or

$$h(c_s) = 4\pi c_s^2 n_s \left(\frac{m_s}{2\pi kT_s}\right)^{\frac{3}{2}} e^{-\frac{mc_s^2}{2kT_s}}$$

The most probable velocity magnitude follows from,



The other (related) definition is when particles are grouped by energies

$$E = \frac{mc_s^2}{2} \rightarrow c_s = \sqrt{\frac{2E}{m_s}} \text{ and } dc_s = \frac{dE}{\sqrt{2m_sE}}$$

In this case,

$$g(E)dE \equiv fd^{3}c_{s} = f4\pi c_{s}^{2}dc_{s} = f\frac{4\sqrt{2\pi}}{m^{3/2}}E^{\frac{1}{2}}dE$$

and so,

$$g(E) = \frac{4\sqrt{2}\pi}{m^{3/2}}E^{\frac{1}{2}}f(E)$$

and for a Maxwellian,

$$f(E) = n_s \left(\frac{m_s}{2\pi kT_s}\right)^{\frac{3}{2}} e^{-\frac{E}{kT_s}}$$



we obtain,

$$g(E) = \frac{2n_s}{\sqrt{\pi}} \frac{E^{\frac{1}{2}}}{(kT_s)^{\frac{3}{2}}} e^{-\frac{E}{kT_s}}$$

The most probable energy follows from,

$$\frac{d \ln g}{dE} = \frac{1}{2E} - \frac{1}{kT_s} = 0 \quad \boxed{E_{\text{most prob.}} = \frac{kT_s}{2}} \quad , \quad (c_s)_{\text{most prob. energy}} = \sqrt{\frac{kT_s}{m_s}}$$

All these velocities are comparable to the speed of sound,

$$c_{\text{sound}} = \sqrt{\gamma \frac{kT_s}{m_s}} = \sqrt{\frac{5}{3} \frac{kT_s}{m_s}}$$
 (for a monoatomic gas)



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