20.310J, 2.797J, 3.053J, 6.024J

MOLECULAR, CELLULAR, & TISSUE BIOMECHANICS

SPRING 2015 GENERAL COURSE INFORMATION

Textbooks and Reference Materials:

Most of the material will come from journal articles and notes to be handed out by the instructors or available through the course website. There are no required textbooks. Texts in the library that are used as sources for material distributed as PDFs include:

- (a) A Grodzinsky, Fields, Forces and Flows in Biological Systems, 2011.
- (b) D. Boal, Mechanics of the Cell, 2001.
- (c) H. Lodish, D. Baltimore, L. Zipurksy, P. Matsudaira, Molecular Cell Biology, 2002.
- (d) K. Dill and S. Bromberg, Molecular Driving Forces, 2003.
- (e) J. Howard, Mechanics of Motor Proteins and the Cytoskeleton, 2001. ISBN-10: 0878933336
- (f) R. Phillips, J. Kondey, J. Theriot, Physical Biology of the Cell, 2008.
- (g) Jackson, M.B. Molecular and Cellular Biophysics. Cambridge University Press (2006). ISBN-10: 0521624703

Texts that may be useful as background material include:

- (h) P. Nelson, Biophysics, 2008.
- M. Mofrad and R. Kamm, Cytoskeletal Mechanics, Cambridge University Press, 2006. ISBN 978-0-521-64828-9 (paperback).
- (j) M. Mofrad and R. Kamm, Cellular Mechanotransduction, Cambridge University Press, 2009.
- (k) Haynie, D. Biological Thermodynamics. Cambridge University Press (2008). ISBN-10: 0521711347
- Wales, D.J. Energy Landscapes. With applications to clusters, biomolecules, and glasses. Cambridge University Press (2004). ISBN-10: 0521814154
- (m) Downarowicz, T. Entropy in Dynamical Systems. Cambridge University Press (2011).
- (n) Ben-Naim, A. Entropy Demystified. World Scientific (2008).

Grading

The term grade will be a weighted average of exams, term paper and homework grades. The general weighting distribution will be

- 60% for the three quizzes (two evening exams and one final exam; 20% each)
- 20% for the term paper (including written report and in-class presentation)
- 20% for the homework

Homework grading is intended to show you how well you are progressing in learning the course material. You are encouraged to seek advice or help from other students and/or to work in study groups. However, all of the work that is turned in must be your own. Please review MIT's academic honesty standards and policies at http://web.mit.edu/academicintegrity/. The homework exercise should be viewed as a learning experience, not a competition. Homework assignments are due in class at the beginning of the lecture on the assigned date.

Course Project and Term paper guidelines

Important dates:

Quiz #1 Quiz #2 Term paper Group presentations Quiz #3 3/10 **evening** 4/14 **evening** 5/8 5/5, 5/7, ?? Final exam week, TBA

20.310/2.797/3.053/6.024 Spring 2015 SYLLABUS Lecture: Tue/Thurs 1-2:30pm, Room 4-270 MOLECULAR, CELLULAR, & TISSUE BIOMECHANICS

Molecular Mechanics

2/3 L#1: Length, Time, & Molecular-scale Forces in Biology (RDK)

Overview of mechanics across all scales. Learning objectives for the term. Scales of force, displacement, time, and energy relevant to biological structures. Molecular forces and bond energies; kT as an energy ruler. Examples drawn from DNA folding, peptide assembly, protein binding, cytoskeletal strain, and tissue compliance. **Readings:** Mahadevan Chapter, Phillips 1.2, 2.1 – 2.3, 3.1, 3.4

2/5 L#2: Single molecule mechanics (RDK)

<u>Thermal forces and Brownian motion</u>. Diffusion, viscous drag, <u>shear</u> forces at low Reynolds numbers. Statistical mechanics and entropy; rubber elasticity and freely jointed chain; Applications to Intracellular Cytoskeleton: rigidity of actin and microtubules.

Readings: Dill & Bromberg 17, Howard 4, 6

MOLECULAR MECHANICS

Biomolecules and intermolecular forces Single molecule biopolymer mechanics Formation and dissolution of bonds Motion at the molecular/macromolecular level

TISSUE MECHANICS

Molecular structure --> physical properties Continuum, elastic models (stress, strain, constitutive laws) Viscoelasticity Poroelasticity Electrochemical effects on tissue properties

CELLULAR MECHANICS

Structure/function/properties of the cell Biomembranes The cytoskeleton Cell adhesion and aggregation Cell migration

Mechanotransduction

Biomechanics at all length scales

biomechanics Networks and Large-scale, Continuum Quantum Molecular **Brownian** discrete or mechanics dynamics mechanics dynamics lumped systems Bone Flight Molecular motors Migration Cartilage Swimming Cytoskeletal rheology Mechanotransduction Cardiovascular Gain analysis system proteins organelles atoms cells organisms organs 10-10 10-9 10-6 10-2 100 meters Courtesy of Eugene Archer on Flickr. License CC BY-NC. Courtesy of Wenging Xu and RCSB Protein

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Courtesy of The Journal of Cell Biology. License: CC BY-NC-SA.

Biomechanics at all length scales

20.310

	Quantum mechanics	Molecular dynamics	Networks and Brownian dynamics	Continuum mechanics	Large-scale, discrete or lumped systems
				Bone Flight	
	Molecula	r motors	Migration Cartilage		Swimming
	Mechanotransduction		Cytoskeletal rheolog	^{JY} Cardiovascula system	r Gain analysis
	atoms	proteins	organelles c	ells organs	organisms
	10-10	^o 10 ⁻⁹	10-6	10 ⁻²	10 ⁰ motors
Cou	rtesy of Wenging Xu an	d RCSB Protein	JCB	Courtesy of Eugene Archer on Flickr. License	

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Biomechanics in the news

Question of the day

ARTICLE

Stimulus-triggered fate conversion of somatic cells into pluripotency

Haruko Obokata^{1,2,3}, Teruhiko Wakayama³†, Yoshiki Sasai⁴, Koji Kojima¹, Martin P. Vacanti^{1,5}, Hitoshi Niwa⁶, Masayuki Yamato⁷ & Charles A. Vacanti¹



Courtesy of Macmillan Publishers Limited. Used with permission. Source: Obokata, Haruko, Teruhiko Wakayama, et al. "Stimulus-triggered Fate Conversion of Somatic Cells into Pluripotency." *Nature* 505, no. 7485 (2014): 641-7.

Martin Karplus 2013 Nobel Prize in Chemistry



Courtesy of Bengt Nyman. License: CC BY.

The <u>Nobel website</u> said the prize was awarded for the researchers' work in "the development of multiscale models for complex chemical systems."

Some Learning Objectives

- 1. To understand the fundamental concepts of mechanics and be able to apply them to simple problems in the deformation of continuous media
- 2. To understand the underlying basis for the mechanical properties of molecules, cells and tissues
- 3. To be able to model biological materials using methods appropriate over diverse length scales
- 4. To be familiar with the wide spectrum of measurement techniques that are currently used to determine mechanical properties
- 5. To appreciate the close interconnections between mechanics and biology/chemistry of living systems

Length scale bars $(10^{-10} \text{ m to } 10^7 \text{ m})$:

Human hair diameter	C. elegans worm length	Eukaryotic cell nucleus diameter	
DNA basepair length	Red blood cell diameter	Microtubule diameter	
E. coli width	HIV diameter	Human genome length	
Human height	Average protein radius	Lipid bilayer thickness	

Time scale bars (1 fs to 4 billion years):

Human lifespan	Stem cell replication	Protein side chain rotation	
Earth age	Covalent bond vibration	Fly lifespan	
Protein folding	Cell crawls 50 um	Sequoia lifetime	

Energy scale bars (1 kT to 10³¹ kT):

AA battery	Sugar (1 g)	Covalent bond	
H-bond	Food you eat today	E. coli replication (1 cell)	
Lightning bolt	Human heart beat (1)	Gasoline (1 gal)	

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How much force does a heart muscle cell generate?

Peak twitch force/cell: $F_{max} = 1.06 \pm 0.20 \mu N$

Force per cross-sectional area: $\sigma = 2.91 \pm 0.65 \text{ mN/mm}^2$



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Source: Yasuda, So-Ichiro, et al. "A Novel Method to Study Contraction Characteristics of a Single Cardiac Myocyte using Carbon Fibers." *American Journal of Physiology-Heart and Circulatory Physiology* 281, no. 3 (2001): H1442-6.

Yasuda, S.-I. et al. *Am J Physiol Heart Circ Physiol* 2001

How much does a cell weigh? $F_{weight} = \rho \times V \times g$ (neglecting the buoyancy force!)

Assume for cell:

- $\rho \sim (1.1x)$ density of water, 1100 kg/m³
- $V \sim 20 \times 20 \times 20 \ \mu m = (20 \times 10^{-6} \ m)^3$
- $g = 9.81 \text{ m/s}^2$

=> F ~ 100 × 10⁻¹² N = 100 pN

Forces at the molecular level

Type of Force	Example	Rupture Force*	Length	Energy
Breaking of a covalent bond	C-C	~1600 pN	0.1-0.5 nm	~ 1.6 × 10 ⁻¹⁹ J ~ 90-350 <i>k</i> T
Breaking of a noncovalent bond.	Biotin/streptavidin	~5-160 pN	~ 1 nm	~ 1.6 × 10 ⁻¹⁹ J ~40 kT
Breaking of a weak bond.	Hydrogen bond	~4-20 pN	~ 0.3nm	~ 4-8 × 10 ⁻²¹ J ~ 1 <i>k</i> T
Developed by molecular motor	Kinesin walking on microtubule	~5 pN	8 nm (step size)	~ 40 × 10 ⁻²¹ Nm ~10 <i>k</i> T

1 nm = 10^{-9} m 1pN = 1 × 10^{-12} N [Energy] = F L = [N m] or [J] k_B = 1.38 × 10^{-23} J/K k_B T = 4.14 × 10^{-21} J = 4.14 pN * 1 nm

* Rupture forces generally depend on the rate of force application! Faster force application requires higher rupture forces!

Relevant molecules in biology

What's inside a cell?

A lot of water, ions (*e.g.* Ca²⁺, Mg²⁺), and lots of small carbon-containing molecules from these four major classes of biological molecules.

Carbohydrate Lipid Protein Nucleic acid

Entropic forces

Entropy:

- $S = k_B \ln (W)$
- k_B = Boltzmann constant
- W = multiplicity, i.e. number of microstates



Review (and a look ahead)

- Random walks and diffusion (see, e.g., Dill & Bromberg)
- Boltzmann statistics

$$P_i = \frac{1}{Q} \exp\left(\frac{-G_i}{k_B T}\right) \qquad Q = \sum_i \exp\left(\frac{-G_i}{k_B T}\right)$$

(note use of free energy, G, as opposed to internal energy, U, corresponding to an ensemble of states, taking entropy into account; G = H - TS)

- Thermal energy, $k_B T = 4 \text{ pN} \cdot \text{nm} = 0.6 \text{ kcal/mole}$ (for T = 300K)
- Persistence length (thermal energy bending stiffness)

$$l_p = \frac{K_b}{k_B T} \qquad K_b = EI$$

More on this later!

RESEARCH



CELL ADHESION

The minimal cadherin-catenin complex binds to actin filaments under force

Craig D. Buckley,¹* Jiongyi Tan,²* Karen L. Anderson,³ Dorit Hanein,³ Niels Volkmann,³ William I. Weis,^{2,4,5}[†] W. James Nelson,^{5,6}[†] Alexander R. Dunn^{1,2,7}[†]

Figures removed due to copyright restrictions. See figure in the research article summary, and figures 2A and 5C in the research article Buckley, Craig D. "The Minimal Cadherin-catenin Complex Binds to Actin Filaments Under Force." *6FLHQFH* 346, no. 6209 (2014): 1254211.

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