## Numerics: Flames are usually solved using Operator Splitting

• Many Equations of form

 $\partial Y_n / \partial t = \text{Reaction}_n (Y) + \text{Transport}_n [Y]$ 

- Typical  $Y_n(x,y,z,t)$  represent mass fraction of *n*th species. 100 species at 10,000 mesh points =  $10^6$  state variables.
- Seldom possible to provide good enough initial guess for steadystate, and hard to solve Newton steps with >10<sup>6</sup> unknowns, so usually time-march to steady-state.
  - Time-marching can be very slow if you need small  $\Delta t!$
- Reaction term is local and very stiff. Transport involves gradients (nonlocal after discretization).
- Usually Chemistry Split from Transport
  - Chemistry solved in parallel using stiff ODE solver (e.g. DSL48S)
  - Transport solved using specialized PDE techniques

# **Operator Splitting Methods**

- Suppose dy/dt = R + T
- Simplest: first integrate dy/dt=T, then start from y(t<sub>f</sub>) and integrate dy/dt=R. Not very accurate.
- Strang: half-step of T, full step R, half step T
  - Second-order accurate, stable.
  - Can be slow to converge to steady-state solution.
- Balanced: dy/dt = (R+c) + (T-c) how to choose c?
  "Simple Balanced": c=1/2 (R(y<sub>n</sub>)-T(y<sub>n</sub>))
- "Rebalanced": use averages of R and T over their steps to get a higher-order implicit formula.
  - more accurate and more stable.
  - see Speth et al. SIAM J. Numer. Anal. (2013)

## Split each timestep into 3 substeps



Conventional (Strang) Splitting applied to a toy problem. Poor behavior near steady state.



(a) Simple Balanced Splitting, h = 0.5

Near Steady State, dY/dt = (Reaction) + (Transport) ~ 0 But each term separately is large. Splitting them makes you walk away from true trajectory during substeps. Balanced Splitting drastically reduces this walk-away.

### Balanced Splitting's Error Exponentially Goes to Zero as Flame steadies out





#### Strang Splitting: Constant Splitting Error

### Rebalanced Splitting: Error Vanishes at steady-state

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10.34 Numerical Methods Applied to Chemical Engineering Fall 2015

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