Problem Set #6, 5.12 Spring 2003 Due Monday, March 31, 4pm

1. a) Rank the following series of molecules based on reactivity in an S_N^2 reaction (Nal/acetone). (1= fastest S_N^2 , 4= slowest S_N^2)



b) Rank the following series of molecules based on reactivity in an S_N1 reaction (EtOH/heat). (1= fastest S_N1 , 4= slowest S_N1)



c) Rank the following series of molecules based on reactivity in an E2 reaction (NaOⁱPr/ ⁱPrOH).
(1= fastest E2, 4= slowest E2)



2. a) Predict the product of the following reaction, and provide the mechanism for its formation.



b) Draw a picture of the transition state for the above reaction. Pay attention to stereochemistry!

c) If the same molecule were heated in diisopropylamine (^{*i*}Pr₂NH), would you expect the same product that you drew in part **a**? Why or why not?

3. Predict the products of the following **substitution** reactions, and specify whether each proceeds by S_N1 or S_N2 . Pay attention to stereochemistry.



4. The reaction conditions are very important in determining what products are obtained in elimination reactions. Predict the **major** products of the following **elimination** reactions.



5. (Bromomethyl)cyclohexane undergoes the following two reactions:



a) Provide a detailed mechanism for reaction (1).

b) Provide a detailed mechanism for reaction (2).

c) Draw reaction-energy diagrams for the reactions on the previous page. Clearly label starting materials, intermediates, products, and the rate-determining step for each reaction.



Reaction Coordinate



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6. The mechanism of the following reaction involves mechanistic steps that you are already familiar with. Provide a detailed mechanism.

