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# Appendix 1b Gaussian 03W on Dell XPS

# Density Functional Theory Calculation of Isotropic Fermi Contact Couplings of Alkyl Semibenzoquinones

(Courtesy of Dr. Mircea Gheorghiu. Used with permission.)

This appendix describes the use of Gaussian 03W Density Functional Theory (DFT) software to calculate hyperfine coupling constants (the isotropic Fermi contact couplings) for semi-1,4-benzoquinone, methyl-, ethyl-, *i*-propyl- and *t*-butyl semibenzoquinones.

Content:

- 1. Building benzosemiquinone structures.
- 2. Calculation of spin-density of semiquinones by DFT.
- 3. Single input file for multiple jobs.
- 4. The output file.

#### 1. Building Benzosemiquinone Structures

The **Gaussian 03W** DFT calculation requires an initial structure that is contained in cartesian coordinates (in Å) or in an input file. The structure can be built interactively with the **GaussView**.

To start GaussView, click on "Shortcut to gview."



In GaussView, there are several windows windows design to assist you in building the geometry for the input file. For example, the main GaussView (see Fig.1) control panel that contains the menu bar, a large variety of toolbars and the Current Fragment window.

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Current Fragment:						

The active **ViewWindow**, in our case **G1:M1:V1**, will immediately contain the benzosemiquinone:



Basically to build the semi-1,4-benzoquinone, you have to start from benzene, add two OH groups, and than strip off the hydrogens. The charge and the spin will be inputed later.



1. Click on the icon with a benzene ring on it. The Select Ring Fragment appears.



2. Next click anywhere in the G1:M1:V1 window. The benzene ring is displayed.



3. Add the oxygen atoms. Double click on the <sup>6</sup>C button. Click on "O" than select at the bottom of the periodic Table the sp<sup>3</sup> bivalent "oxygen" fragment (S-S-LP-LP).

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<u>File E</u> dit ⊻iew	Select Element
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🕨 Current F	Na Mg Al Si P S CI Ar
	K CaSc Ti V Cr Mn Fe Co Ni Cu Zn GaGe As Se Br Kr
	RbSrYZrNbMoTcRuRhPdAgCdInSnSbTeIXe
	CsBaLaHfTaWReOsIrPtAuHgTIPbBiPoAtRn
	Fr Ra Ac Rf Db Sg Bh Hs Mt
	CePrNdPmSmEuGdTbDyHoErTmYbLu
	Th PaUNpPuAmCmBkCfEsFmMdNoLr
	Select Oxygen Fragment:
	Atom

4. Click on any hydrogen from the benzene ring, an **OH** will be inserted. Click on the hydrogen from the *para* position. A second **OH** will be attached.

B	<b>本 今</b>	B
	Oxygen Tetravalent (S-S-LP-LP)	• •
	🖪 G1:M1:V1 - New	
	•	.ton
	Build Select Placement	

5. Click on **Delete Atom** button (**left to the ? button**) and delete the two hydrogen atoms attached to the **OH** groups.



Screenshot images courtesy of Gaussian, Inc. Used with permission. ESR Appendix 1b-5 6. The benzosemiquinone built so far is not necessarily symmetrized  $D_{2h}$ . Under **Edit** click on the **Point Group...**.



For example, the **G1:M2 Point Group Symmetry** window appears. Check the box **Enable Point Group Symmetry**. From this window you will learn that your molecule



has C1 symmetry, however you can convert this geometry to  $D_{2h}$  by clicking **Symmetrize**. If you do so you will get a  $D_{2h}$  benzosemiquinone.

Enable Point Group Symr	netry	Curren	nt point group	D2h
Constrain to subgroup:	C1	~	For:	All changes 💌
Approximate higher-order po	oint groups:	-		
Tolerar	nce: Def	ault (0.01)	~	Symmetrize
	10		1	
Alexandra data	ouromotru			

Click **OK**. The current point group is now  $D_{2h}$ . Now you are ready to start the calculation.

### 2. Calculation of Spin-Density of Alkylsemiquinones by DFT

At this stage, you have to enter the setup to set the calculation parameters and define the level of theory used.

1. Click on Calculate (GaussView 3.08), then on Gaussian....



2. Gaussian Calculation Setup window will appear.

b Type Metho	d Title	Link 0	General	Guess	NBO	PBC	Solvation	
otimization 💌								
imize to a	Mir	iimum 💌	] 🗖 U:	se GDIIS				
culate Force Cor	stants Ne	ver 🗸	ι — Ο Πυ:	se tiaht cor	iveraence	e criteria		

- The Job Type is optimization
- In Method choose ground state, DFT..., Unrestricted, B3LYP. Leave the Basis Set as 3-21G. It will be edited latter to epr-II (this latter basis set is not an option available from this window). Change Charge to -1 and Spin to doublet.
- In **Title** window type the a **Name for your job**.
- **Solvation** should be **None**.
- In Additional keywords, add the keywords density=current.
- Now we want to make extra changes to the input file, which is being written with these parameters. Click on edit... button. Save the file in d:/5.33 directory.
- A notepad file will pop-up. Now is time for the final editing:

%chk=semiqui %mem=6MW %nproc=1	inone-mg.chk				
#T opt ub313	/p/epr_II geom=cc	nnectivity	density=c	urrent	
benzosemiqui	inone D2h				
-1 2					
с					
с	1	B1			
с	2	B2	1	Al	
С	3	вз	2	A2	1
С	4	В4	3	A3	2
С	1	B5	2	A4	з
н	1	B6	6	A5	5
н	2	В7	1	A6	6
н	4	B8	3	A7	2
н	5	В9	4	A8	з
0	3	B10	2	A9	1
0	6	B11	1	A10	2

- 3. Type a "T" after the pound sign to make **#T** instead of **#** (**T** is keyword for terse printing).
- 4. Replace **ub3lyp/3-21g** (or any other basis set) with **ub3lyp/epr-II**.

For the time being, please ignore the first three lines. For the time being, the first significant is the fourth line, the so-called route line, which begins with the keyword "#". Here "T" is keyword for terse output printing. The keyword "opt" requests optimization of the geometry of molecule. Electronic structure of the radicals (open shell system) are calculated in the unrestricted model ( $\alpha$  electrons and  $\beta$  electrons are handled separately). The keyword "u" prepending the method keyword, requests open shell calculation. The method of calculation, density functional theory, is "b3lyp" that stands for Bethe-style-3 parameter DFT using the Lee-Yang-Parr correlation functional. The keyword for the basis set used in the calculation is "epr-II". This basis set was designed by Barone<sup>1</sup> and are optimize for the computation of the <u>hyperfine coupling constants</u> by DFT methods. Other keywords will be available consulting the main reference that will be on the computer Table.

<sup>&</sup>lt;sup>1</sup> Barone, V. in *Recent Advances in Density Functional Methods, Part I,* Chong, D. P., Ed., World Scientific Publ. Co.: Singapore, 1966.

- 5. Save the file (File) and Exit.
- 6. The **RunGaussian** window appear. Click on **Okay** button.

ds: 'Mult.:	# op -1 2	ot ub3lyp	geom=con	nectivity e	pr-ii dens	ity=curre	ent
,ре	Metho	od Title	Link 0	General	Guess	NBO	PBC
point	Save:	None Don't sav	/e 🔁 Ru	n Gaussia			
		2)	Submit	the following 3\semiquinor	∣ file to Gau ne-mg.gjf	issian?	
				<u>0</u> k	<u>C</u> anc	el	

The calculation will start immediately. Details regarding the actual processing is displayed in the **GaussView 3.08** window.



Screenshot images courtesy of Gaussian, Inc. Used with permission.

To complete the run strating from the builder geometry, it takes ca. 10 minutes for benzosemiquinone  $D_{2h}$ , ca. 2 hours for methylbenzosemiquinone, ca. 5.5 hours for

ethylbenzosemiquinone, ca 9 hours for the *i*-propylbenzosemiquinone and ca 15 hours for t-butylsemibenzoquinone. Budget your time wisely. Sign up for calculation as early as possible.

You can run a multiple job file. See the next discussion (Single input file for multiple jobs).

## 3. Single Input File for Multiple Jobs.

The input file to submit, for example, three calculations is displayed next in blue (on the web site) Berlin Sans FB fonts to differ from the regular text. Suggestions:

- 1. Start with the benzosemiquinone file (perhaps saved under a different name).
- 2. Build methylbenzosemiquinone. Save the input file.
- 3. Leave a line after the last information regarding the benzosemiquinone. Type in (correctly) the link line, the check file name, and the route line.
- 4. Leave a line as space.
- 5. Open the file with the second molecule. Copy and paste the title, charge, multiplicity and the geometric information concerning the second molecule in the multi jobs input file.
- 6. For the third (next molecule) go to step 1 through step 5.
- 7. If the new molecule is the last input, add the NoSave command (see the example below).
- 8. Leave to empty lines at the end.

%chk=semiquinone.chk %mem=6MW %nproc=1 #T opt ub3lyp geom=connectivity epr-ii density=current

benzosemiquinone D2h

-1 2 insert the geometry of benzosemiquinone

--link1--%chk=semiquinone.chk #T opt ub3lyp geom=connectivity epr-ii density=current

#### methyl benzosemiquinone

-1 2 insert the geometry of methylbenzosemiquinone

--link1--%chk=semiquinone.chk %NoSave # opt ub3lyp geom=connectivity epr-ii density=current

t-butyl benzosemiquinone

# -1 2 insert the geometry of t-butylbenzosemiquinone

## 4. The Output File

The output (or LOG) file for the b3lyp/epr-II calculation will provide you with the isotropic Fermi contact couplings. This file should be saved with the same name as your input file with a .LOG extension. In the figure below is an example of the semi-benzoquinone output file. The couplings are found in column 5 in values of Gauss. Here, because all hydrogens are equivalent (group  $D_{2h}$ ), the coupling constant value is equal at -2.39 Gauss. Also information regarding the optimized geometry of the anion radical is also available.

Symmetry	/ B2U	KE= 4.49125042278	5D+01		
Jynnicer y	, 000	Isotro	pic Fermi Contact	t Couplinas	
A	Atom	a.u	. MegaHert:	z Gauss	10(-4) cm-1
1 0	:(13)	-0.000	42 -0.47178	-0.16834	-0.15737
2 0	:(13)	-0.000	42 -0.47178	-0.16834	-0.15737
3 0	:(13)	-0.008	95 -10.05990	-3.58962	-3.35562
4 0	:(13)	-0.000	42 -0.47178	-0.16834	-0.15737
5 C	:(13)	-0.000	42 -0.47178	-0.16834	-0.15737
6 0	:(13)	-0.008	95 -10.05990	-3.58962	-3.35562
7 F	H(1)	-0.001	.50 -6.70870	-2.39383	-2.23778
8 H	H(1)	-0.001	.50 -6.70870	-2.39383	-2.23778
9 H	+(1)	-0.001	.50 -6.70870	-2.39383	-2.23778
10 H	+(1)	-0.001	.50 -6.70870	-2.39383	-2.23778
11 0	(17)	0.035	57 -21.56538	-7.69506	-7.19344
12 0	(17)	0.035	57 -21.56538	-7.69506	-7.19344

All calculations in Gaussian03W end with a proverb. Here is a superb sample:

13.893 0.0000 0.0000 1.0000 ваа -0.5756 41.651 14.862 13.682 -27.575 12 0(17) врр -0.5669 41.018 14.636 0.0000 1.0000 0.0000 1.1425 -82.669 -29.498 1.0000 0.0000 BCC 0.0000 \_\_\_\_\_ 1|1|UNPC-UNK|FOPt|UB3LYP|EPR-II|C6H402(1-,2)|PCUSER|16-Sep-2003|0||# 0
PT UB3LYP GEOM=CONNECTIVITY EPR-II DENSITY=CURRENT||benzosemiquinone D
2h||-1,2|C,-1.2300173355,0.,-0.6900845267|C,-1.2300173355,0.,0.6900845
267|C,0.,0.1.4758944703|C,1.2300173355,0.,0.6900845267|C,1.2300173355
,0.,-0.6900845267|C,0.,0.,-1.4758944703|H,-2.1641832335,0.,-1.251621652
2|H,-2.1641832335,0.,-1.2516216522|H,2.1641832335,0.,1.2516216522|H,2.
1641832335,0.,-1.2516216522|0,0.,0.,2.7469855669|0,0.,0.,-2.7469855669
||Version=x86-Win32-G03RevB.04|state=2-B2G|HF=-381.5936519|s2=0.761767
|s2-1=0.|s2A=0.750042|RMSD=6.115e-009|RMSF=1.179e-004|Dipole=0.,0.,0.|
PG=D02H [C2"(01C1.C101),sG(C4H4)]|@ IT IS A SIMPLE TASK TO MAKE THINGS COMPLEX, BUT A COMPLEX TASK TO MAKE THEM SIMPLE. Job cpu time: 0 days 0 hours 9 minutes 52.0 seconds. File lengths (MBytes): RWF= 44 Int= 0 D2E= 0 Chl Normal termination of Gaussian 03 at Tue Sep 16 14:10:17 2003. 0 Chk= 9 Scr= 1

Table. The calculated hyperfine coupling constant,  $a_{H}$ , (in gauss), according to equation 10 and your data.



Benzosemiquinone		
R is the substituent	DTF: B3LYP/EPR-II	Your data
R=H		
$H_2$		

Benzosemiquinone		
R is the substituent	DTF: B3LYP/EPR-II	Your data
R=Me		
$H_2$		
$H_3$		
$H_4$		
average CH <sub>3</sub>		
R= isopropyl		
H <sub>2</sub>		
H <sub>3</sub>		
$H_4$		
СН		
average CH <sub>3</sub>		