5.80 Small-Molecule Spectroscopy and Dynamics Fall 2008

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<u>Lecture #32</u>: The $H_2CO \widetilde{A}^1A_2 \leftarrow \widetilde{X}^1A_1$ Transition

H₂CO was 1st (asymmetric top) polyatomic electronic transition to be rotationally analyzed.

G. H. Dieke and G. B. Kistiakowsky Phys. Rev. 45, 4 (1934).

It is more complicated than linear HCCH because many values of the K_a rotational quantum number can have significant thermal population in the $\sum_{n=0}^{n} = 0$ level. For a linear molecule in a Σ -state, $K_a = 0$, and $\ell = 0$ in $\sum_{n=0}^{\infty} = 0$.

The $S_1 \leftarrow S_0$ transition in H_2CO

- * is electronically forbidden in $C_{2v}(a)$ [(x, y, z) = (c, b a)]
- * the excited state is expected and appears to be non planar hence C_{2v} may not be relevant.
- * "quasi-planar" molecule inversion barrier is low resulting in staggering of bending levels
- * 3 distinct transition mechanisms, each with its own selection rules, contribute to $\tilde{A} \tilde{X}$ system.

Outline:

- i. classification of orbitals and normal modes
- ii. what do we expect (geometry and vibrational structure of S_1)
- iii. "vibronically" rather than electronically allowed system false origin, promoter mod
- iv. surprise in hot band spectrum peculiar spacings in upper state out-of-plane bend
- v. low barrier to inversion through planarity

don)

Body fixed axis system:



Molecular orbitals from atomic orbitals:

- IP CO 14.014eV
- IP H 13.595 ∴ H atom 1s orbitals lie above CO HOMO
- IP C 11.264
- IP O 13.614 $\therefore \sigma 2s, \pi 2p$ are polarized toward O



Figure 1: Correlation of the orbitals of planar H₂XY to those of the united molecule Y₂ and to those of 2H + XY. The variable along the abscissa is the XH distance. Note that at the left, since Y₂ is homonuclear, the orbitals are σ_g , σ_u , π_g , π_u while at the right, since XY is heteronuclear, the *g*, *u* characteristic does not strictly apply. However, just as at the left, the orbitals σ_{2s} are mixtures (but not 50:50 mixtures) of the 2*s* orbitals of X and Y and similarly for the other XY orbitals. The order of $\pi_{(u)}2p$ and $\sigma_{(g)}2p$ is reversed at the right compared to the left in accordance with the situation in CO as compared to O₂ (see Herzberg, Vol. I, p. 346). At the left, the splitting of π_u and π_g into b₂ (π in-plane) and b₁ ($\pi \perp$ plane) corresponds to breaking the cylindrical symmetry of the C + H + H = O united atom.





The C-O antibonding orbital, viewed edge-on, is the LUMO: B₁ [odd wrt C₂, even wrt $\sigma_v(xz)$]

 $\begin{array}{c} HH & \bigoplus \\ & paper \ is \ xz \ plane \\ & HOMO \rightarrow LUMO \ is \ B_1 \leftarrow B_2 \qquad \pi^* \leftarrow n \\ & S_1 \ state \ configuration \ is \ \dots \ b_1^1 b_2^1 \end{array}$

 $B_1 \otimes B_2 = A_2$ which is electric dipole forbidden from the $\widetilde{X}^1 A_1$ electronic ground state (because A_2 does not transform as T_x , T_y , or T_z).

What do we know about the $\widetilde{X}^1 A_1$ ground state?

Figure out symmetries of vibrations from $\Gamma_{\text{RED}} = [12, -2, 2, 4] = 4\Gamma_{A_1} + 1\Gamma_{A_2} + 3\Gamma_{B_1} + 4\Gamma_{B_2}$

Vibrations: $3 \times A_1$, $0A_2$, $1B_1$, $2B_2$ (after removing 3 translations and 3 rotations)

 $2e^{-}$ in each orbital $\widetilde{X}^{1}A_{1}$

C₀ 1.1343

nan and negative m	ertial defect implies planali	(y)
$\omega_1(a_1)$	2766.4 cm^{-1}	sym CH stretch
$\omega_2(a_1)$	1746.1	CO stretch
$\omega_3(a_1)$	1500.6	scissors
$\omega_4(b_1)$	1167.3	out of plane
$\omega_5(b_2)$	2843.4	anti-sym CH
$\omega_6(b_2)$	1251.2	in plane wag

Numbering of vibrational modes, by convention in order of frequency within each symmetry type, symmetries in order A_1 , A_2 , B_1 , B_2 .

What do we expect for $\pi^* \leftarrow n$ electronic transition, provided that it is observable?

- * lengthen C–O bond
- * change hybridization on C from sp² to sp³ to minimize antibonding interaction! (Could also think of this in a Walsh diagram sense, out of plane distortion allows H atoms to derive some bonding character from $\pi^*(b_1)$ and thereby polarize it out of the antibonding C–O region.)

This could cause $\begin{cases} \theta_{\text{HCH}} \rightarrow 109^{\circ} \text{ (decrease from 121.1}^{\circ}) \\ \text{nonplanar distortion} \end{cases}$

If the molecule becomes non-planar, we can no longer work in the C_{2v} point group. The vibrations no longer divide into 3 symmetry species. In C_s we have



Mode 4 (out of plane bend) could become symmetry active in all quanta, not merely even quanta! If C_s , expect progression in v'_4 . So whether we see odd and even quanta of v'_4 seems to be a key question.

Strongly F–C active modes in C_s symmetry:

 $\begin{array}{lll} \mbox{definitely} & CO & \omega_2 \\ \mbox{possibly} & scissors & \omega_3 \\ & & & & & \\ & & & & & \\ \mbox{possibly active because of } sp^2 \rightarrow sp^3 \mbox{ change of hybridization on } C) \\ \mbox{possibly} & \mbox{out-of-plane} & \omega_4 \end{array}$

Observe long progression in 1182 cm⁻¹: must be C–O stretch. (Note that $\pi^* \leftarrow$ n decreases bond order from 2.0 to 1.5.) Each strong band is observed with short progressions built on it:

in 824 cm⁻¹ out -of-plane and 1322 cm⁻¹ scissors and 2872 cm⁻¹ symmetric CH $\theta_{\rm HCH} = 121.1^{\circ}$

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But is 824 cm⁻¹ ω'_4 or $2\omega'_4$? (Why would we even think of $2\omega'_4$? Seems too low a frequency.)

HINT:
$$\begin{cases} A = 8.69 & B = 1.156 & C = 1.041 & \text{for excited state} \\ \frac{1}{A} + \frac{1}{B} - \frac{1}{C} = 0.0195 \text{ cm: much larger (and positive!) inertial defect than in } \widetilde{X} \text{ state.} \end{cases}$$

suggests much less planar than $\widetilde{X}^1 A_1$ state, so if equilibrium geometry is nonplanar $\rightarrow C_s$ point group $\rightarrow 824 \text{ cm}^{-1}$ is $1 \omega'_4$ not $2 \omega'_4$. (In C_s , there are 4A' symmetry allowed normal modes and two A" forbidden normal modes. The fourth A' mode would be an out of plane bend.) In C_{2v} the ${}^1A_2 \leftarrow {}^1A_1$ transition is electronically forbidden, but it might be vibronically allowed.

$$\Gamma_{(ev)'} \otimes \Gamma_{(ev)''} = \Gamma_a \text{ or } \Gamma_b \text{ or } \Gamma_c$$

Start with cold bands $\Gamma_{(ev)''} = \Gamma_{A_1}$.

In
$$C_{2v}$$
, expect to only find strong transitions built on odd quanta of non-totally symmetric modes:
 $v'_4(b_1) \Rightarrow \Gamma_{(ev)'} = A_2 \otimes B_1 = \Gamma_{B_2} = \Gamma_b$ (b-type)
 v'_5 or $v'_6(b_2) \Rightarrow \Gamma_{(ev)'} = A_2 \otimes B_2 = B_1 = \Gamma_{B_1} = \Gamma_c$ (c-type)
rotational selection rules

<u>Starting from V'' = 0</u>:

So expect to see $\mathbf{V'} = (0 \text{ or } 1, v_2, 0 \text{ or } 1, 1, 0, 0)$ b-type bands from $\widetilde{X}(0, 0, 0, 0, 0, 0)$ level. Never expect to see $\mathbf{V'} = (0, 0, 0, 0, 0, 0)$ i.e. the 0_0^0 band.

Modes 5 and 6 are not expected to be F-C active. (antisymmetric CH stretch, in-plane wag)

All strong <u>cold bands</u> in $\widetilde{A} - \widetilde{X}$ spectrum are observed to be b-type ($\Delta K_a = \pm 1, K_c = \text{odd}$).

So all looks clear for a non-planar ${}^{1}A' S_{1}$ electronic state OR a vibronically allowed ${}^{1}A_{2} \leftarrow {}^{1}A_{1}$. But is 824 cm⁻¹ v'_{4} or 2v'_{4}? It would have to be 2v'_{4} if the S_{1} state is planar (C_{2v}), ${}^{1}A_{2}$. It could be v₄ if the S₁ state is nonplanar (C_s)¹A'.

Hot bands and emission bands contain some surprises!

Level diagram - for hot bands

 v_4'' is lowest frequency mode in \widetilde{X} -state.



These vibrational bands have some contribution from the rotational constants A' or A" for K' = 1 or K'' = 1 included, which we will ignore here.

Observed Bending Level Diagram



Why is this so irregular? (It appears that 824 cm⁻¹ is $v'_4 = 3 - v'_4 = 1$ level spacing, implying C_{2v} symmetry!)

$$(A' = 9 \text{ cm}^{-1})$$







get a situation where it is appropriate to work in the higher symmetry group.

J. C. D. Brand, J. Chem. Soc. p. 858 (1956)

deduced that zero point energy in the inversion potential for the \tilde{A} –state is only 400 cm⁻¹, and that the top of the barrier is 650 cm⁻¹, thus the v₄ = 1 level lies just below top of barrier.

What mechanisms could make the $\widetilde{A}^1A_2 - \widetilde{X}^1A_1$ transition observable?

1. Vibronic coupling via the Q₄ out-of-plane vibration: $\begin{cases} see only alternate quanta of v'_4 or v''_4 \\ b-type rotational selection rules \end{cases}$

2. Magnetic dipole
$$(A_1 \otimes A_2) = A_2 = R_z \swarrow z$$
-component of magnetic dipole
a-type transitions

a-type rotational selection rules [actually observed as very weak a-type transitions into the inversion doublet component opposite to that observed via the main bands]

 A_2

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3. Vibronic coupling via the v_5 or v_6 in plane vibrations $\begin{cases}
built on v'_5 = 1 \text{ or } v'_6 = 1 \\
c-type \text{ bands}
\end{cases}$

Weak bands of this type are also observed.

The $\widetilde{A} - \widetilde{X}$ transition does derive its intensity from 3 mechanisms:

1. Vibronic coupling to a ${}^{1}B_{1}$ excited electronic state via the v_{4} (b₁) promoting mode (this is the only B₁ overall vibrational level)

2. Vibronic coupling to a ${}^{1}B_{2}$ excited electronic state via v_{5} and v_{6} (b₂) promoting modes (only B₂ overall vibrational levels)

3. a-type magnetic dipole transitions (only totally symmetric vibrations). All 3 symmetric modes are governed by Franck-Condon like intensity factors.

In the vibronic coupling cases, we have to worry about the vibrational structure of the state from which intensity is borrowed, times a factor controlling the strength of the vibronic coupling which increases monotonically with the number of odd quanta in the promoting mode.

Rotational structure:

 $A_2 has \chi(\sigma_v) = -1 \text{ (odd parity)}$

total parity of A_1 vs. A_2 states



The J_{1J-1} level always lies at higher energy than the J_{1J} level. The b-type vibronic (via the $\nu'_4(b_1)$ promoting mode, admixing 1B_1 electronic character) rotational transition $J_{1J-1} \leftarrow J_{0J} {}^rQ_0(J)$ terminates on

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the upper asymmetry component, whereas the c-type vibronic (via the v'_5 or v'_6 (b₂) promoting mode, admixing ${}^{1}B_2$ electronic character) rotational transition $J_{1J} \leftarrow J_{0J} {}^{r}Q_0(J)$ terminates on the lower asymmetry component. The sign of the observed asymmetry splitting, determined by A, B, C, thus provides a way of distinguishing between the two vibronic coupling mechanisms. For a magnetic dipole allowed transition, parity does not change. An a-type magnetic dipole ${}^{1}A_2 - {}^{1}A_1$ rotational transition

$$(J+1)_{1(J+1)} \leftarrow J_{1J} \qquad \qquad {}^{q}R_{1}(J)$$

is from the lower energy asymmetry component in the lower state to the lower energy asymmetry component in the upper state. The allowed R and P branch magnetic dipole transitions are lower to lower and upper to upper.