

## Atmos. Chem. Lecture 8, 9/30/13: Stratospheric Chemistry 2

- Review: Ozone chemistry
- Bromine (and other halogens)
- Heterogeneous chemistry of NO<sub>y</sub>
  - Polar ozone loss
  - Montreal Protocol

*PSet 3 due Wednesday, Oct 9  
No lecture Wed Oct 2; guest lecture Mon Oct 7*

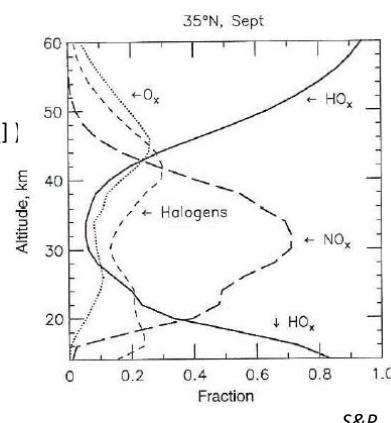
### Review

Chemical families: O<sub>x</sub>, HO<sub>x</sub>, NO<sub>x</sub>, ClO<sub>x</sub> (+ BrO<sub>x</sub>)

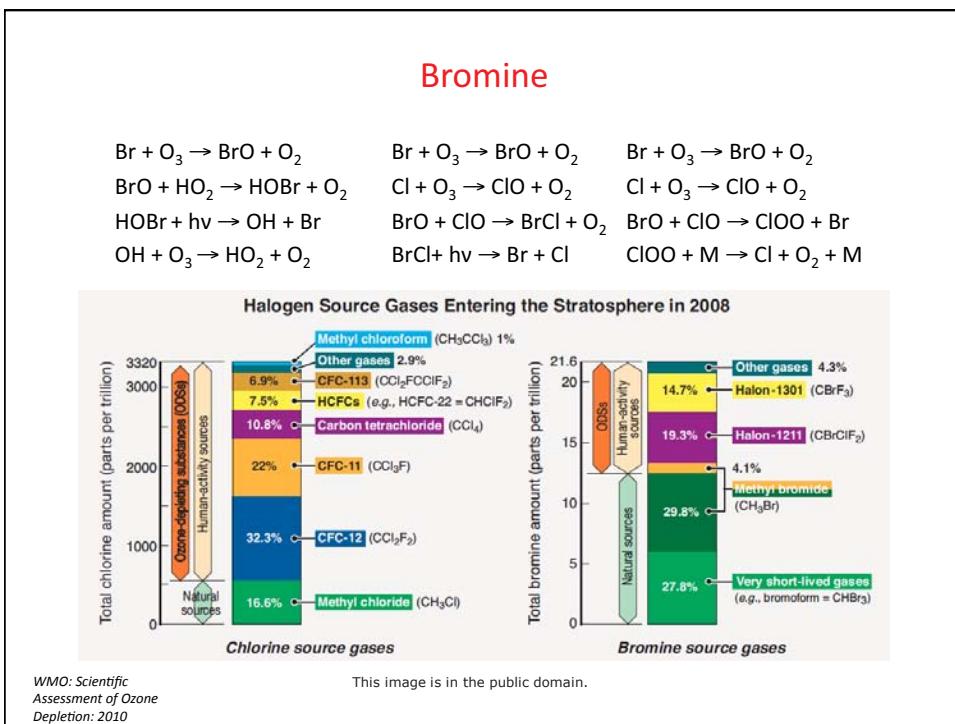
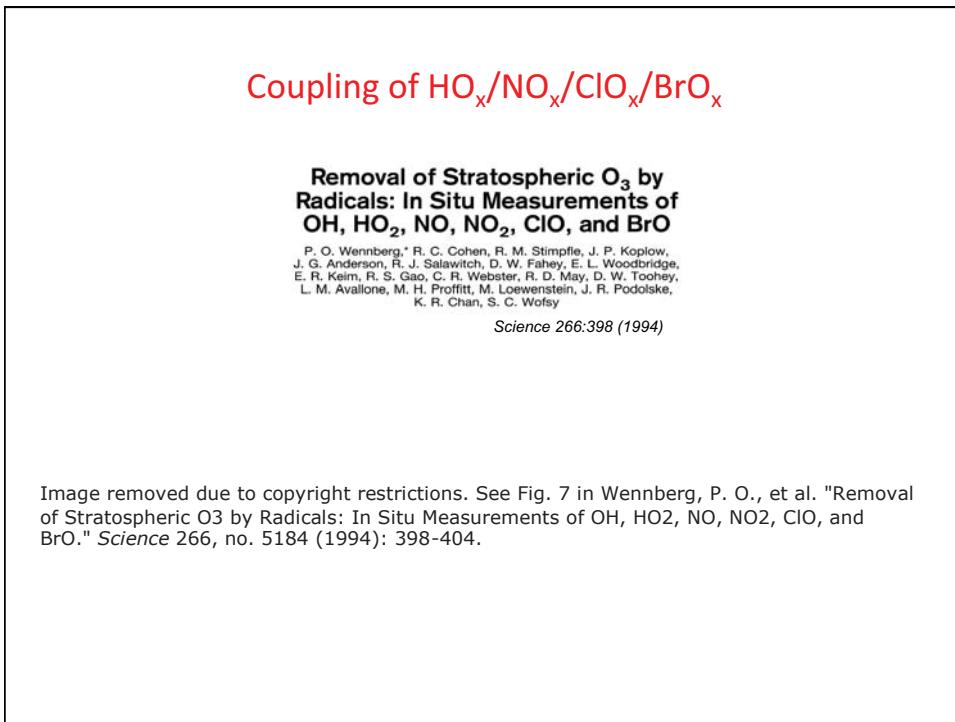
Chemical loss of "odd oxygen" ( [O<sub>x</sub>] = [O<sub>3</sub>] + [O]  $\approx$  [O<sub>3</sub>] )  
by numerous channels:

- direct reaction (O + O<sub>3</sub>)
- catalytic loss by HO<sub>x</sub>, NO<sub>x</sub>, ClO<sub>x</sub> (+ BrO<sub>x</sub>) radicals

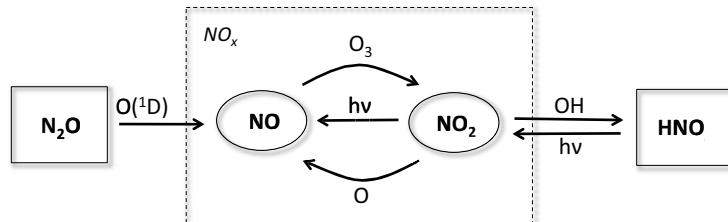
Important couplings between families  
Importance of reservoir species



© John Wiley and Sons. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <http://ocw.mit.edu/help/faq-fair-use/>.

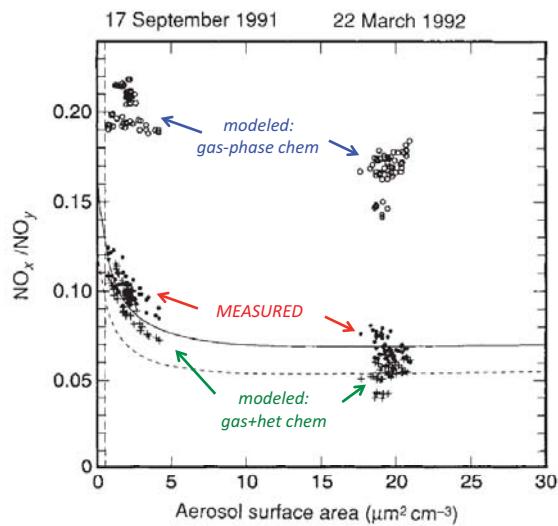


## Revisiting NO<sub>y</sub> chemistry



*[Note: Additional material is discussed here during lecture.]*

## Overestimate of [NO<sub>x</sub>]/[HNO<sub>3</sub>]



© Macmillan Publishers Ltd. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <http://ocw.mit.edu/help/faq-fair-use/>.

Fahey et al. *Nature*  
363:509 (1993)

## Effects of N<sub>2</sub>O<sub>5</sub> hydrolysis

Images removed due to copyright restrictions. See Fig. 15a and 15b in McElroy, M. B., R. J. Salawitch, et al. "The Changing Stratosphere." *Planetary and Space Science* 40, no. 2-3 (1992): 373-401.

McElroy et al., *Planet. Space Sci.*, 40:373 (1992)

## Ozone hole"

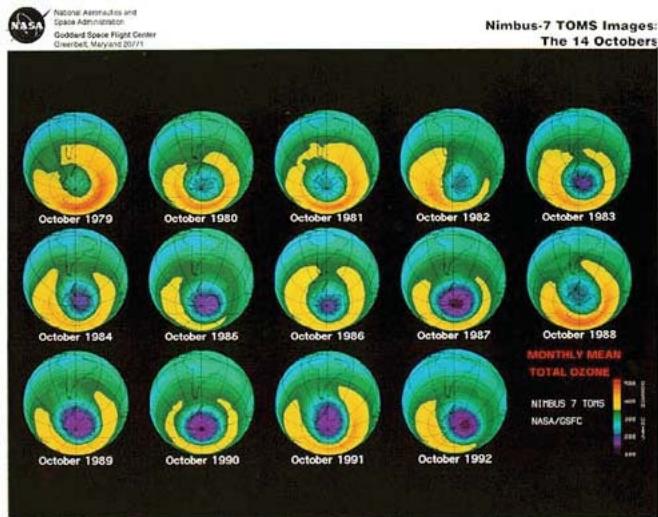
Abstract of the journal article removed  
due to copyright restrictions.

Image removed due to copyright restrictions.  
See Fig. 1 in Jones, A. E., and J. D. Shanklin,  
"Continued Decline of Total Ozone over Halley,  
Antarctica, since 1985." *Nature* 376, no. 6539:  
371-447.

*Nature* 376:409 (1995)

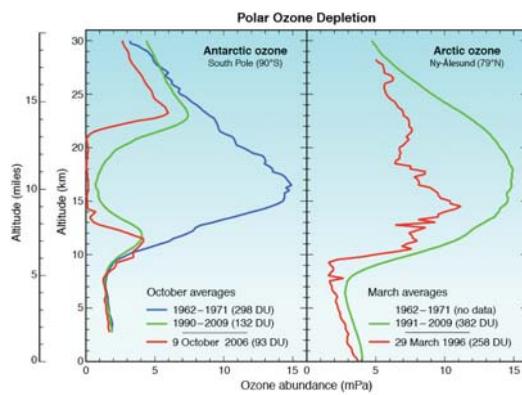
*Nature* 315:207 (1985)

## TOMS (Total Ozone Monitoring Satellite) Data



This image is in the public domain.

## Profile of the “ozone hole”

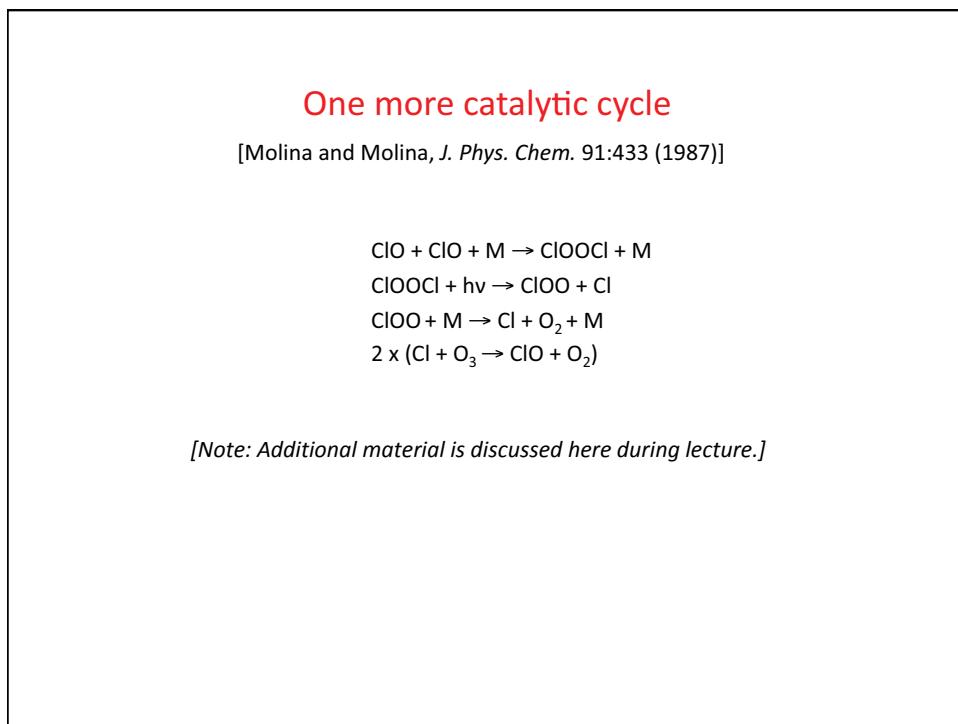
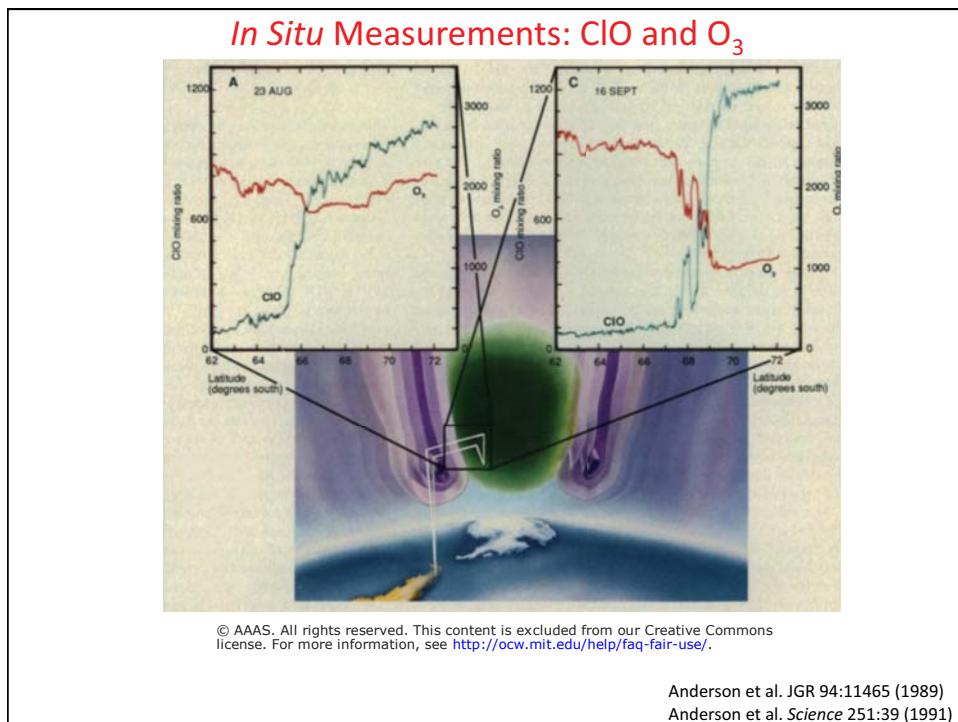


This image is in the public domain.

### *Suggested causes:*

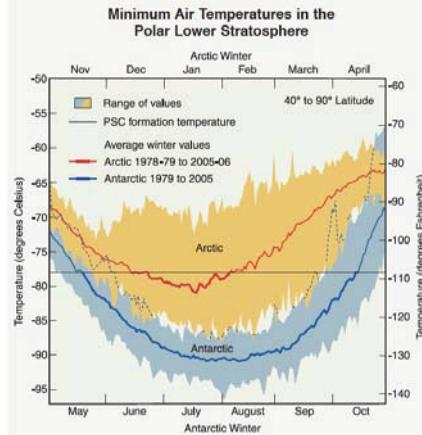
- O<sub>3</sub>-poor tropospheric air
- solar events: high NO<sub>x</sub>
- “Other theories too numerous to mention never made it beyond the popular press.” (*Science* 251:39, 1991)

*WMO: Scientific  
Assessment of Ozone  
Depletion: 2010*



## Polar vortex

Polar night → air from upper stratosphere cools, descends  
 Large temperature gradients with lower latitudes → high westerly winds, seals off area

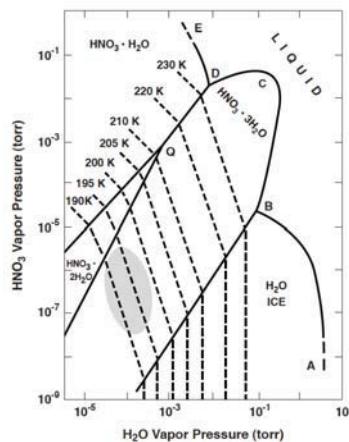


WMO: Scientific  
Assessment of Ozone  
Depletion: 2006

This image is in the public domain.

## Polar Stratospheric Clouds (PSCs)

Binary solutions ( $\text{H}_2\text{O} + \text{HNO}_3$ ) or ternary solutions ( $\text{H}_2\text{O} + \text{HNO}_3 + \text{H}_2\text{SO}_4$ )



- 1)  $\text{HCl}(\text{ads}) + \text{ClONO}_2 \rightarrow \text{Cl}_2 + \text{HNO}_3(\text{ads})$   
 Solomon et al., *Nature* 321:755 (1986)

Other heterogeneous reactions also:  
 $\text{ClONO}_2 + \text{H}_2\text{O}(s) \rightarrow \text{HOCl} + \text{HNO}_3(\text{ads})$

- 2) Gravitational settling  
 Voigt et al., *Science* 290:1756 (2000)  
 Fahey et al., *Science* 291:1026 (2001)

© Princeton University Press. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <http://ocw.mit.edu/help/faq-fair-use/>.

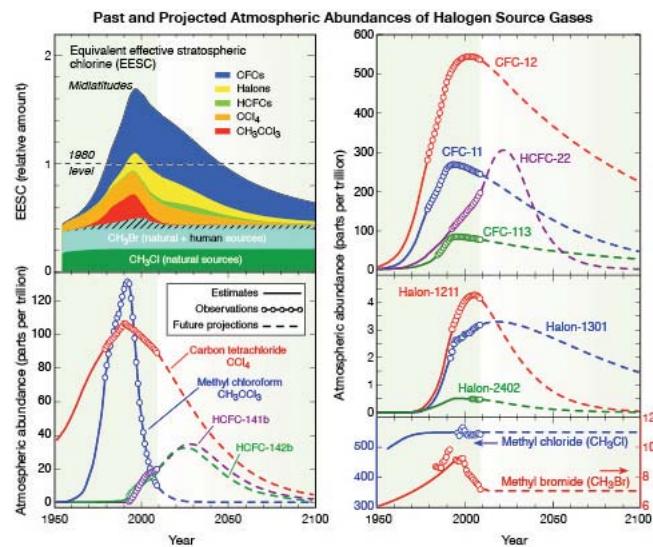
Jacob

## Chronology of the “ozone hole”

Image removed due to copyright restrictions. See Figure 10-13 in Jacob, Daniel. *Introduction to Atmospheric Chemistry*. Princeton University Press, 1999.

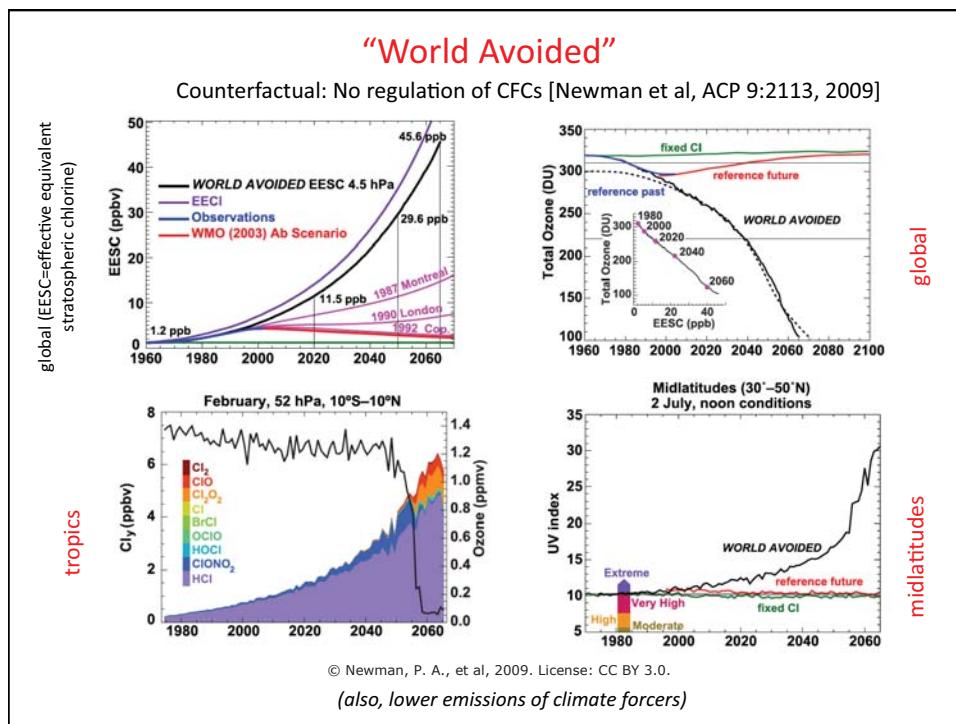
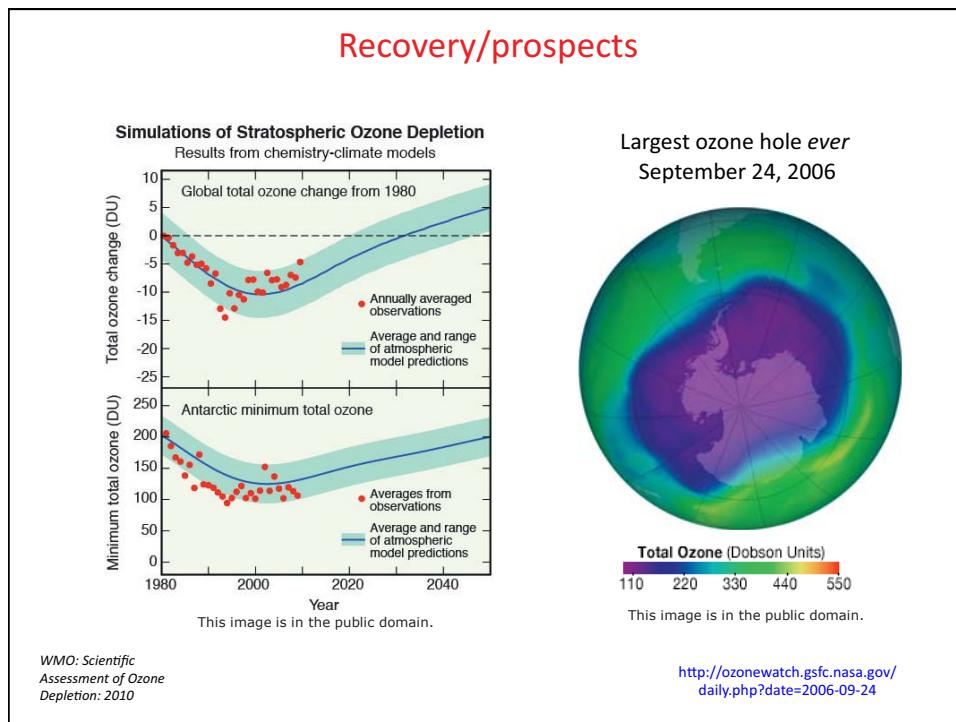
*Jacob*

## Montreal Protocol (1987, w/amendments)



WMO: Scientific  
Assessment of Ozone  
Depletion: 2010

This image is in the public domain.



MIT OpenCourseWare  
<http://ocw.mit.edu>

1.84J / 10.817J / 12.807J Atmospheric Chemistry

Fall 2013

For information about citing these materials or our Terms of Use, visit: <http://ocw.mit.edu/terms>.