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3.21 Kinetics of Materials—Spring 2006

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Lecture 10: Activated Jump Processes.

References

1. Balluffi, Allen, and Carter, Kinetics of Materials, Sections 7.1–7.1.2 and 7.2

Key Concepts

- Diffusing particles in a solid move between neighboring positions of minimum energy, making what are called *jumps*. This is a cooperative process involving not only the jumping particle but also the surrounding ones—hence, the process is *many-bodied* and difficult to describe in detail.
- In *KoM* we describe three models for *thermally activated* jumps of of an interstitial species in a crystal, with gradually increasing complexity: a one-particle model in which the diffusing species sit in a square-well potential, a one-particle model in which the potential is harmonic, and a many-particle model with a harmonic potential. The resulting relations for the interstitial jump frequency Γ' , Eqs. 7.9, 7.13, and 7.25 all have the form $\Gamma' = \nu \exp[-G^m/(kT)]$. The jump rate is equal to the product of an "attempt frequency" and a Boltzmann-Arrhenius exponential factor containing a migration activation free energy.
- Diffusion of a particle involves a series of discrete jumps $\vec{r_i}$ and the total displacement of a given particle is a sum over all the individual jump vectors. The "spreading" of a distribution of jumping particles is quantified by the *mean-square displacement* $\langle R^2(N_\tau) \rangle$ in *KoM* Eq. 7.31. Equation 7.31 is completely general.
- The relation between the mean square displacement and the diffusivity can be obtained from the second moment of the concentration distribution from a point source (see *KoM* Section 7.2.1). In three dimensions, isotropic diffusion for a time τ obeys the relation $\langle R^2(N_{\tau}) \rangle = 6D\tau$.
- If a particle moves by a series of displacements, each of which is independent of the one preceding it, the particle moves by a *random walk*. In crystals, random walks are confined to specific sites in the crystal, and jumps are confined to a specific set of directions.
- KoM Section 7.2.2 presents a random-walk model for diffusion on a one-dimensional lattice that provides an elegant way to connect microscopic jumping processes to macroscopic diffusion by considering the probability distribution arising from a point source in one dimensional random walks. This approach leads to the important relation in one dimension D = Γ⟨r²⟩/2 and in three dimensions the analogous relation is D = Γ⟨r²⟩/6.
- In many important diffusion processes diffusion is not a random walk because successive jumps do not occur at random. The degree of nonrandom jumping is quantified by the *correlation coefficient* f defined in *KoM* Eq. 7.49. f = 1 for a random walk. For diffusion in three dimensions, D = Γ(r²)f/6.

Related Exercises in Kinetics of Materials

Review Exercises 7.1; 7.3–7.6, pp. 159–161.