3.23 Electrical, Optical, and Magnetic Properties of Materials Fall 2007

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3.23 Fall 2007 – Lecture 25

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Exam

- Monday 17, 9am-12noon
- 2 questions on electronics, 1 on optics, 1 on magnetic

PEN(S), CALCULATOR (THAT WORKS) I SHEFT · BAND STRUCTURE (Si, Ge, Ga AS)

Syllabus

- From particles to waves: the Schrödinger equation
- The mechanics of quantum mechanics: operators, expectation values
- Measurements and probabilities. The harmonic oscillator.
- The hydrogen atom and the periodic table
- Periodicity and phonons
- Electrons in a lattice: Bloch's theorem
- The nearly-free electron model
- The tight-binding model. Band structures
- Metals, semiconductors and insulators
- Intrinsic and extrinsic semiconductors
- Transport of heat and electricity
- The p-n diode

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Syllabus

- Electromagnetism in dielectric media
- Classic propagation of waves
- Optical materials and refractive index
- Interband absorption
- Excitons and luminescence
- Fundamental of ferromagnetic materials
- Hysteresis loop and driving energies
- Hard materials and permanent magnets
- Soft materials: thin films and nanoparticles. Spintronics and GMR
- Spin valves, spin switches, and spin tunneling

Last time

- Fermi's golden rule, joint density of states
- Perturbing Hamiltonian, selection rules
- Frequency-dependence of band adsorption in direct or indirect band gap SC. Absorption above the band edge
- Excitons and exciton absorption
- Luminescence: low-carrier density; degeneracy.

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Study

 Fox, Optical Properties of Solids, Chap. 7.5, 9.4, 10.4

Photoluminescence spectroscopy



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Electroluminescence: LED

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Please see: Fig. 5.12 and 5.13 in Fox, Mark. Optical Properties of Solids. Oxford, England: Oxford University Press, 2001.

Diode laser: resonant longitudinal modes



Injection current, threshold gain, slope efficiency



Electroluminescence: diode laser

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Please see: Fig. 5.16 in Fox, Mark. Optical Properties of Solids. Oxford, England: Oxford University Press, 2001.

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Solid state lasers

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Please see: Fig. 9.12 in Fox, Mark. Optical Properties of Solids. Oxford, England: Oxford University Press, 2001.

Plasmons Hecketrons x unit volum.

$$N = \frac{Neu}{\varepsilon_{\pi}}$$
of Solids.

$$Nm \frac{d^{2}u}{dt^{2}} = -Ne\varepsilon$$

$$= -\left(\frac{N^{2}e^{2}}{\varepsilon_{0}}\right)u$$

$$Wp = \left(\frac{Ne^{2}}{\varepsilon_{0}}\right)^{\frac{1}{2}}$$

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Please see: Fig. 7.12 in Fox, Mark. *Optical Properties of Solids*. Oxford, England: Oxford University Press, 2001.

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Plasmonics

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Polarons

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Please see: Fig. 10.8 in Fox, Mark. Optical Properties of Solids. Oxford, England: Oxford University Press, 2001.

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Electron Transfer

- Electron transfer mediated by polar solvent fluctuations.
- Juff Heling of Docur when reactant and product are degenerate

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Please see Fig. 2 in Sit, P. H.-L., et al. "Realistic, Quantitative Descriptions of Electron-tranfer Reactions: Diabatic Free-energy Surfaces from First-principles Molecular Dynamics." arXiv:cond-mat/0606310v1 [cond-mat.mtrl-sci], 2006.



• $\epsilon = \epsilon_{product} - \epsilon_{reactant}$ is the energy gap or reaction coordinate

R. A. Marcus, J. Chem. Phys. 43, 679 (1965) A. Warshel, J. Chem. Phys. 86, 2218 (1982)

D. Chandler, Classical and Quantum Dynamics in Condensed Phase Simulations, pp. 25-49 (1998) 2.23 Electronic, Optical and Magnetic Properties of Materials - Nicola Marzin (MI1, Pall 2007)

Ferrous-ferric self-exchange

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Please see: Fig. 1 in Sit, P. H.-L., et al. "Realistic, Quantitative Descriptions of Electron-tranfer Reactions: Diabatic Free-energy Surfaces from First-principles Molecular Dynamics." arXiv:cond-mat/0606310v1 [cond-mat.mtrl-sci], 2006.

- ΔG is the free-energy barrier of the reaction
- λ (reorganization energy) is the difference in free energy between product and reactant in the optimum atomic configuration for the reactant

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The colour of gems



Fe²¹ Ti^ht

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Please see: Fig. 10.10 and 10.11 in Fox, Mark. Optical Properties of Solids. Oxford, England: Oxford University Press, 2001.