MEASUREMENT in QUANTUM MECHANICS

YOUNG DOUBLE SLIT EXPERIMENT



Ref: A. Tonomura, J. Endo, T. Matsuda, T. Kawasaki and H. Ezawa Am. J. of Phys. 57, 117 (1989)

Electrons are emitted one by one from the source in the electron microscope. They pass through a device called the "electron biprism", which consists of two parallel plates and a fine filament at the center. The filament is thinner than 1 micron (1/1000 mm) in diameter. Electrons are then detected one by one as particles at the detector.

The electrons were accelerated to 50 kV, and therefore their speed is about 40% of the speed of the light (120,000 km/sec). So they pass through a one-meter-long electron microscope in 10⁻⁸s. There is no more than one electron in the microscope at one time, since only 10 electrons are emitted per second. The experiment lasts 20 minutes (video 1 min!)

VIDEO HITACHI RESEARCH LAB. HTTP://WWW.HITACHI.COM/RD/RESEARCH/EM/ DOUBLESLIT.HTML

Each electron position is random...



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... but the probability of its position is predicted by the wavefunction: $P(x) = |(x)|^2$. Measuring the position of many electrons, we learn about average properties, $\langle x \rangle = \int x |(x)|^2 dx$.



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Here, we discover the probability distribution of their positions

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(a) The buckyball carbon-70 (and C-60) (1999) (b) the pancake-shaped biomolecule tetraphenylporphyrin (TPP) $C_{44}H_{30}N_4$; (2003) (c) the fluorinated fullerene $C_{60}F_{48}$. (2004)

TPP is the first-ever biomolecule to show its wave nature. $C_{60}F_{48}$ has an atomic mass of 1632 units and currently holds the world record for the most massive and complex molecule to show interference.

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